




Tritium- Control Technologies for TMSR in CAS

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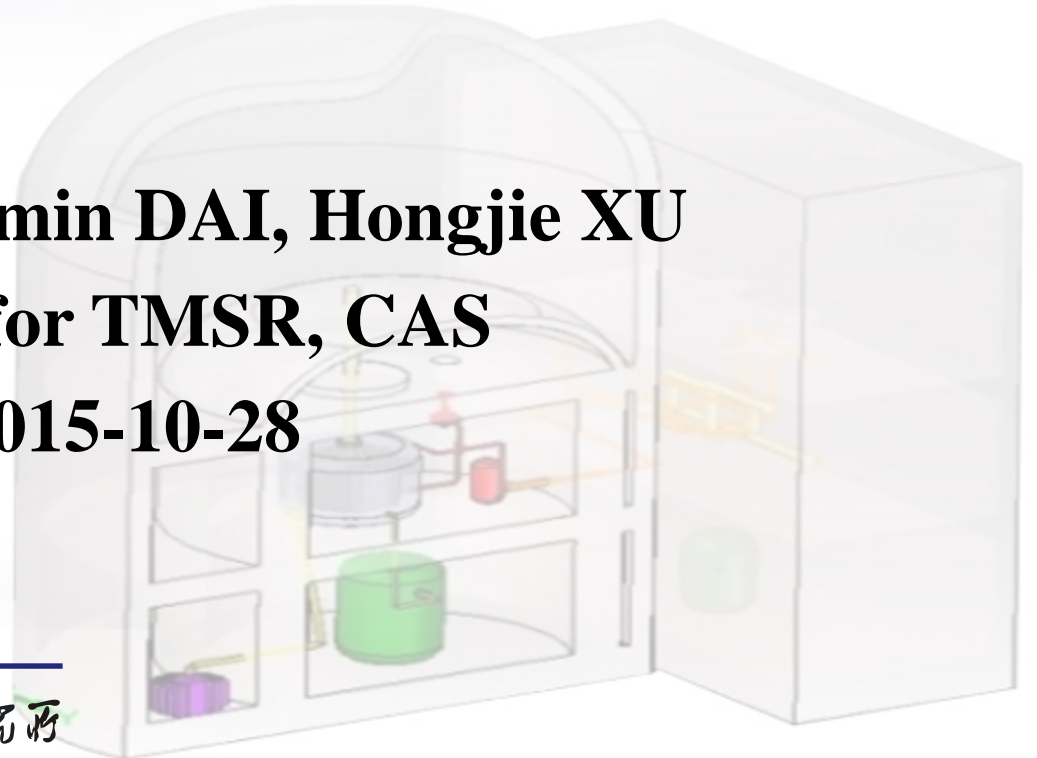


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Shanghai Institute of Applied Physics, Chinese Academy of Sciences



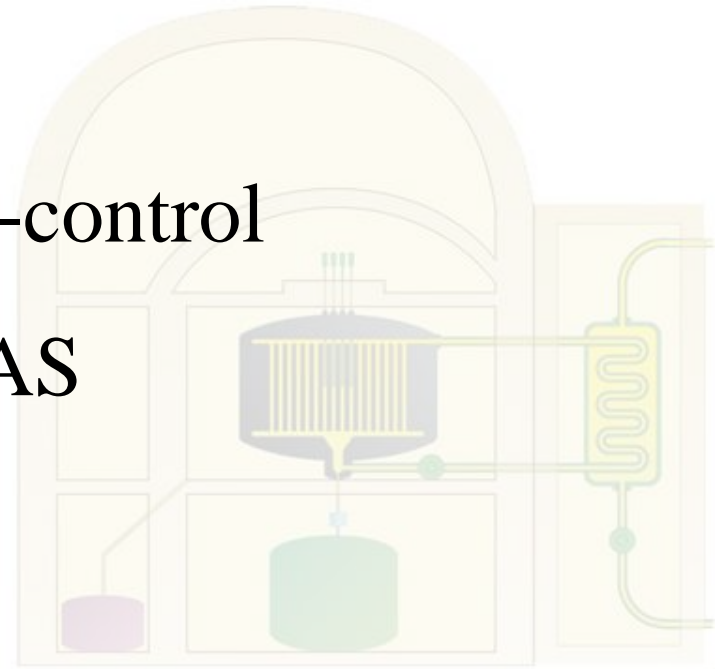
Outline

I. Background

II. Roadmap of Tritium-control technologies for TMSR in CAS

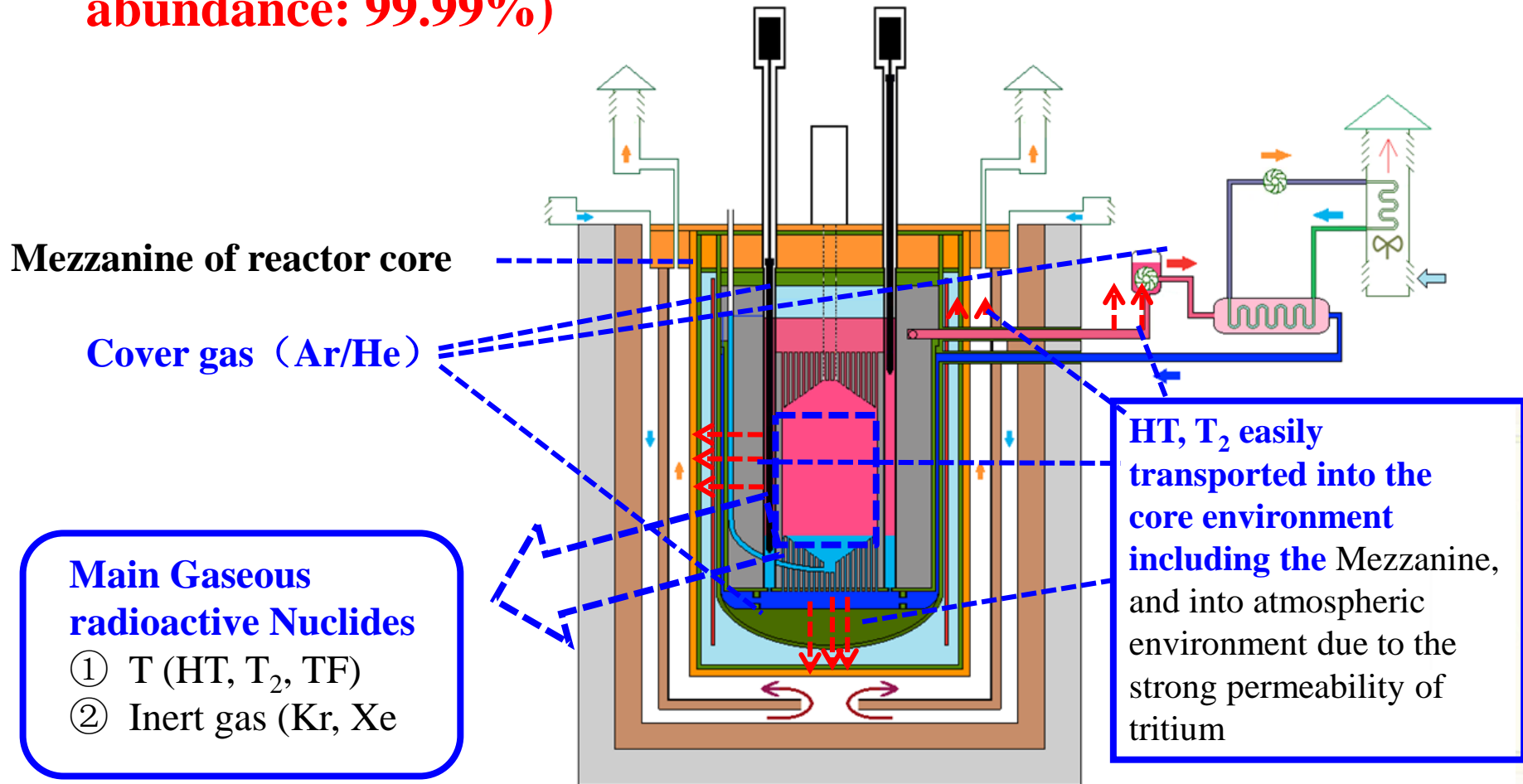
III. Performances at the tritium-control technologies for TMSR in CAS

IV. Summary



I. Background

1. Tritium production in TMSR (FLiBe as primary coolant, ${}^7\text{Li}$ abundance: 99.99%)

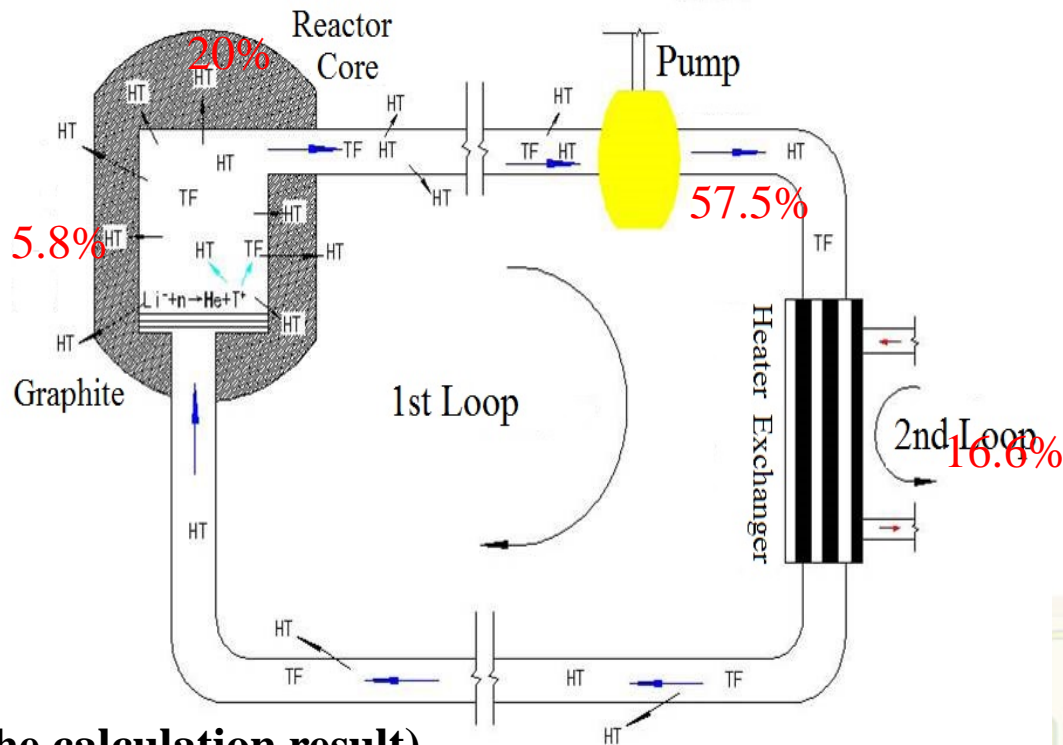


2. Distribution of Tritium in TMSR

Tritium distribution in TMSR-SF1

Position	%	Bq/a*
Core	20 (Cited from MSRE)	$5.5E14$
Graphite	5.8	$1.6E14$
Mezzanine	5.8	$1.6E14$
1 st Loop	57.5	$1.6E15$
2 nd Loop	16.6 (Cited from MSRE)	$4.6E14$
Total	100	$2.8E15$

* $10MW_{th}$, 7Li 99.99%, 300-days run (the calculation result)



Emission limit of tritium in 3000MW nuclear power station in China: $1.5E13$ Bq/a (LWR), $4.5E14$ Bq/a (HLR)

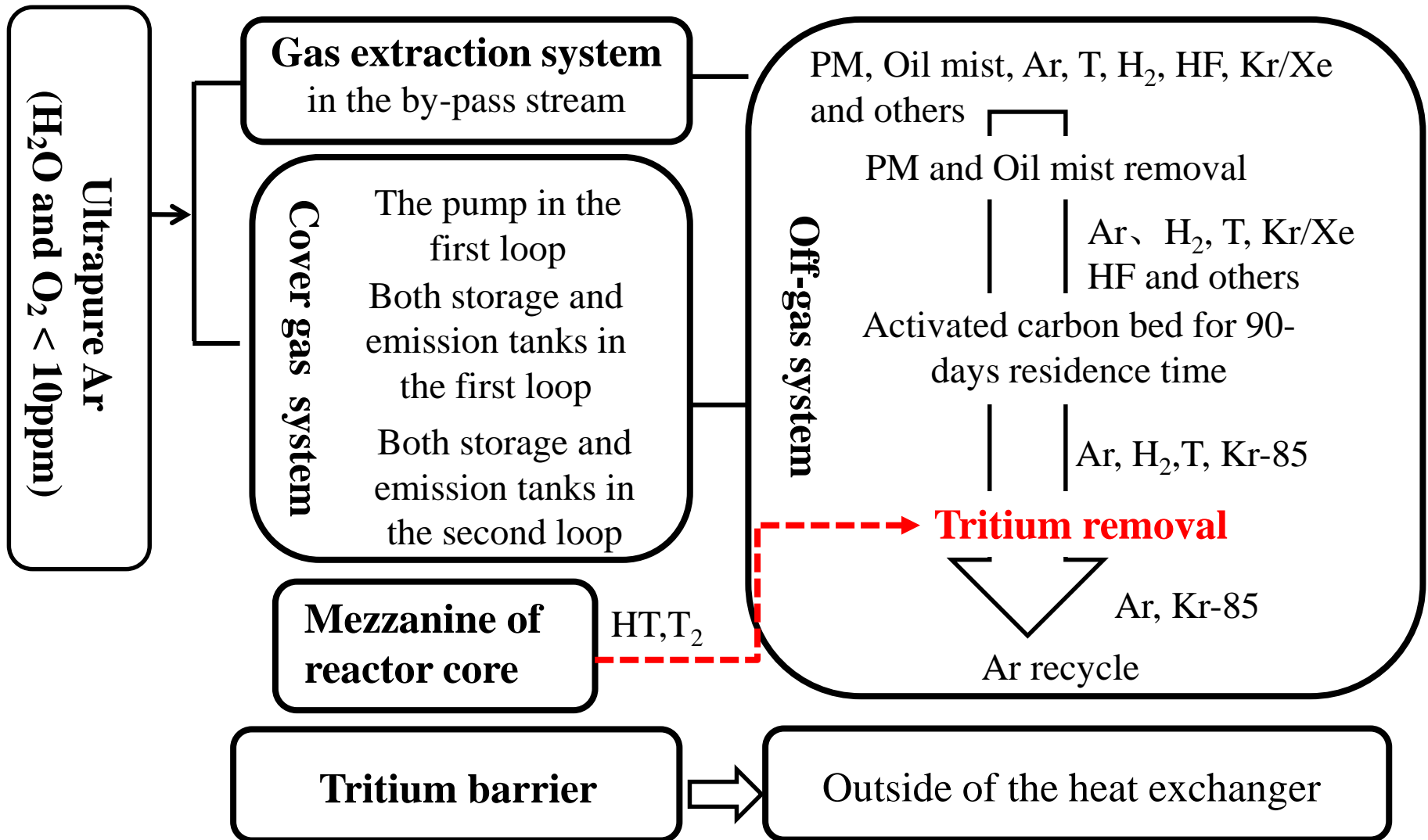
Does tritium transport into the cover gas completely? What is the chemical form of tritium?

Tritium removal: necessary and difficult

3. Considerations of Tritium control for TMSR

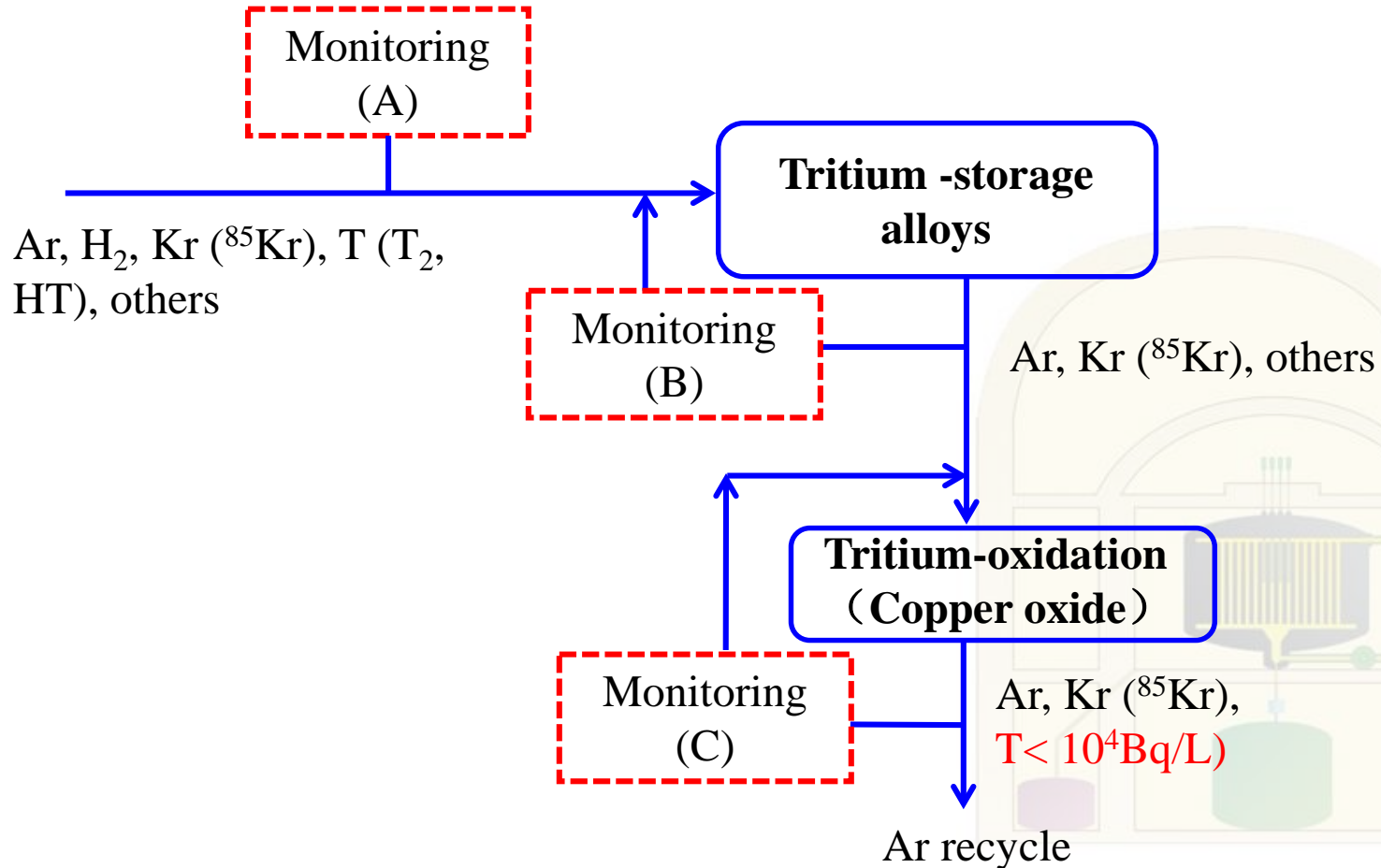
Remove tritium in the first loop as much as possible

Prevent tritium permeation into the second loop as much as possible



Tritium removal

Combination of **tritium -storage alloys** and **tritium-oxidation** with the **on-line tritium monitoring**



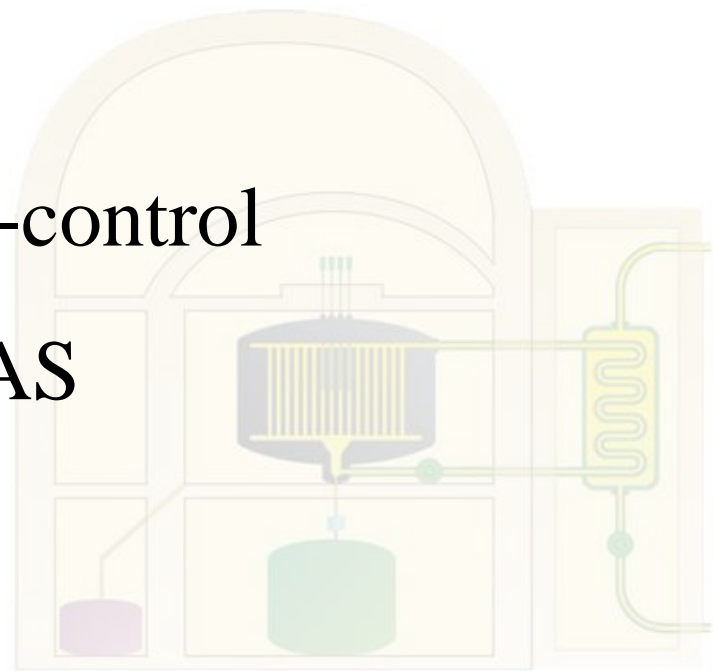
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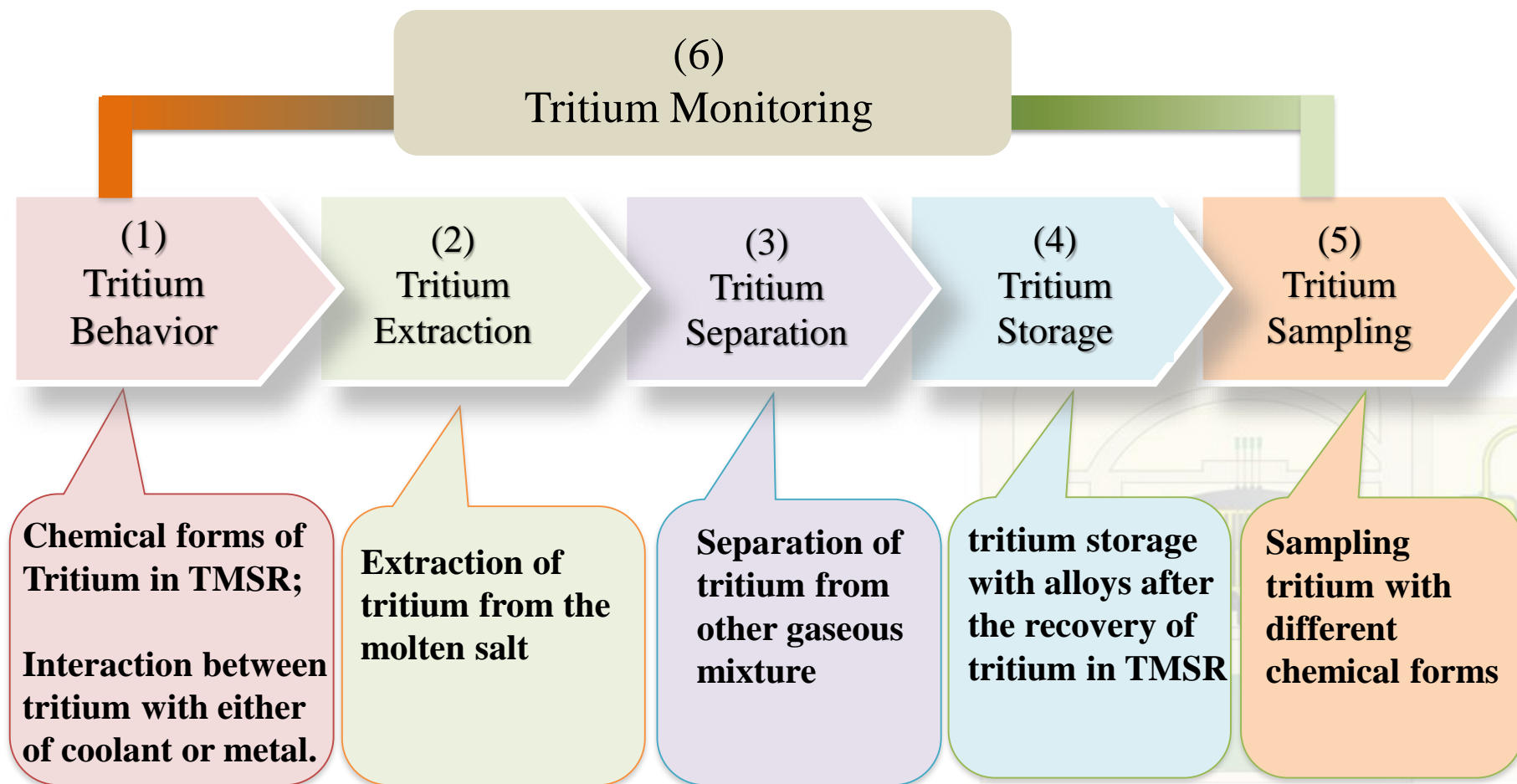
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II. The Roadmap of tritium-control technologies for TMSR in CAS



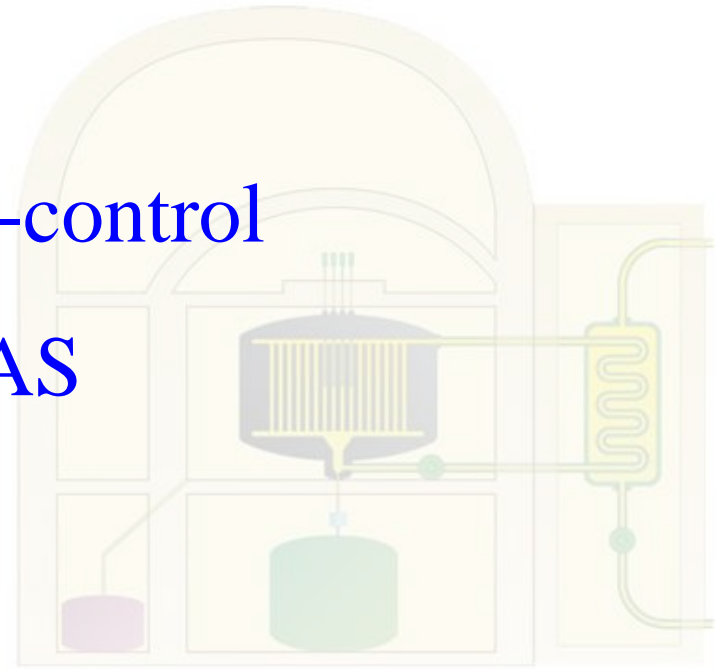
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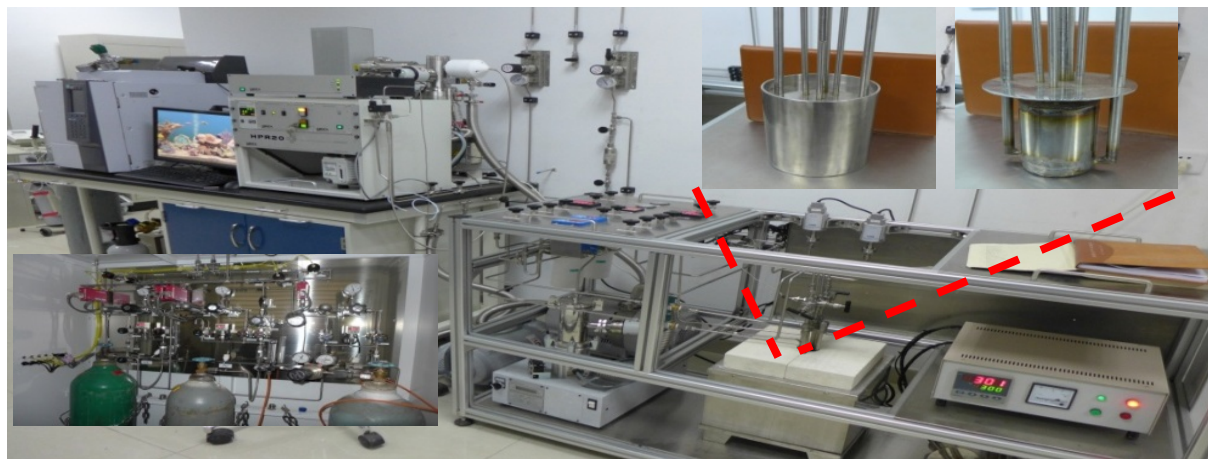
IV. Summary



1. Tritium behavior

Interaction of hydrogen isotopes with either of **high-temperature molten salt** or **structural materials**

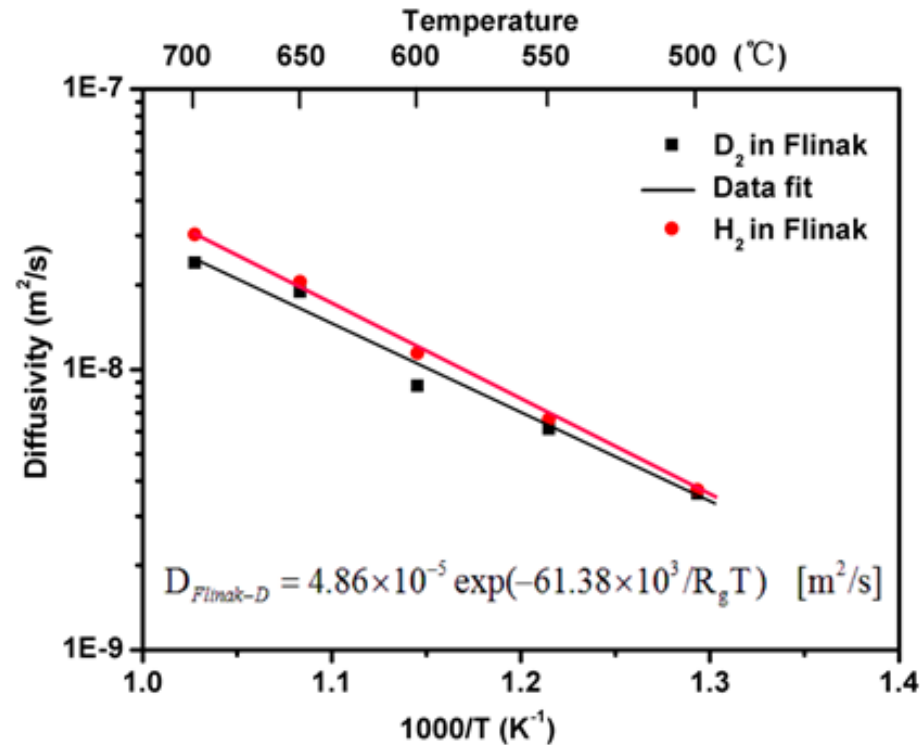
- Test apparatus for diffusion of hydrogen isotopes **in high-temperature molten salt**



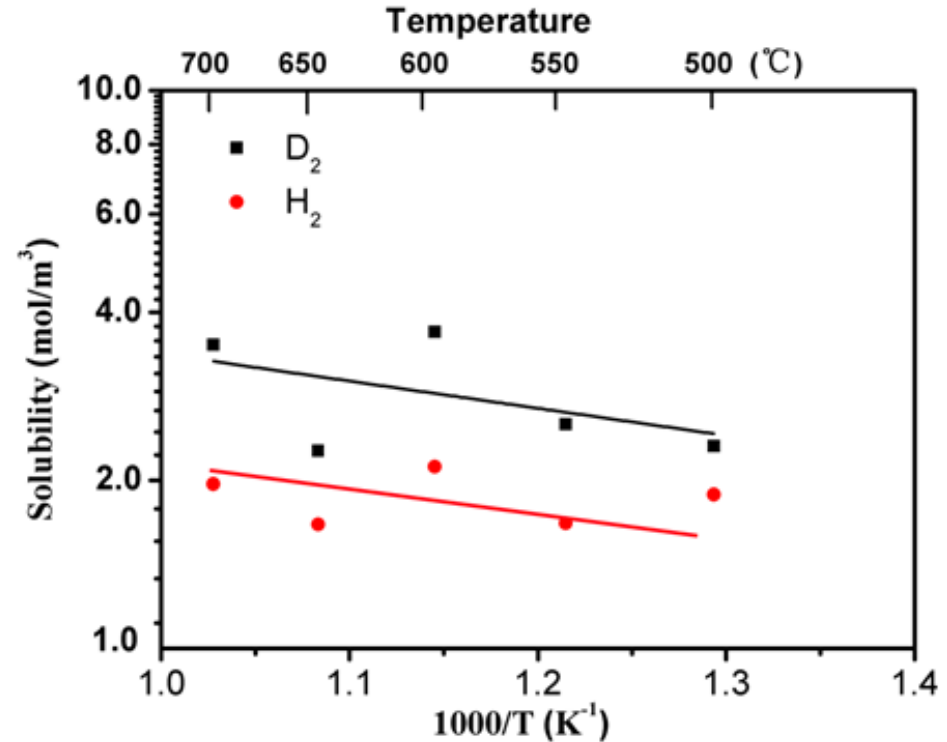
- Test apparatus for permeation of hydrogen isotopes **in structural materials**



➤ Diffusion of Hydrogen isotopes (H_2 , D_2) in high-temperature FLiNaK

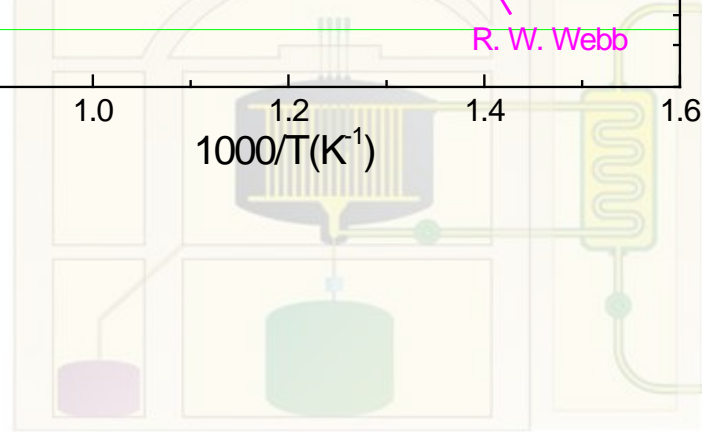
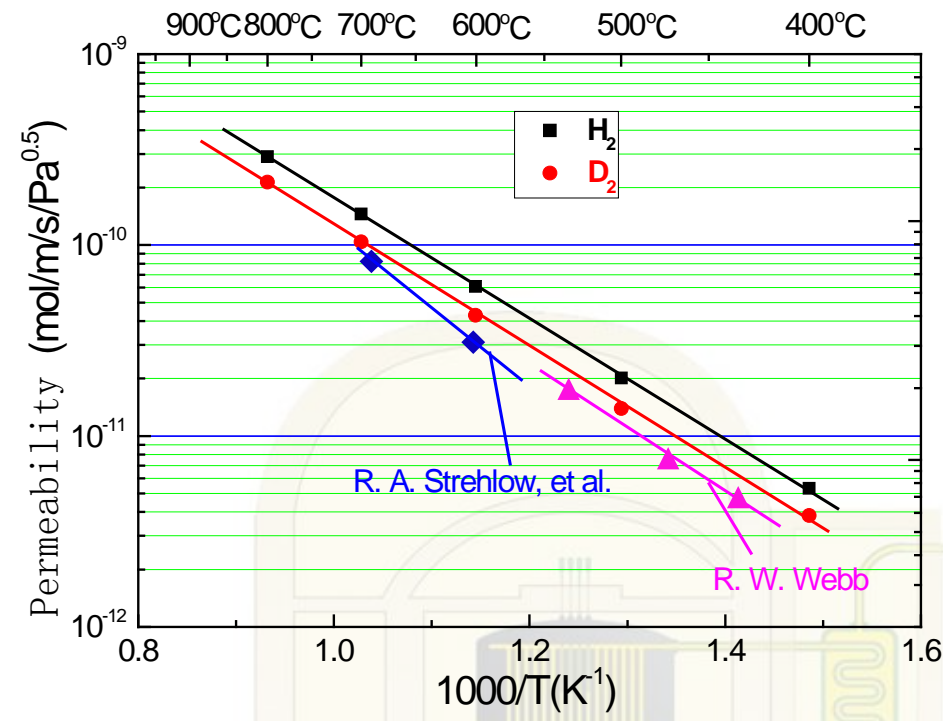
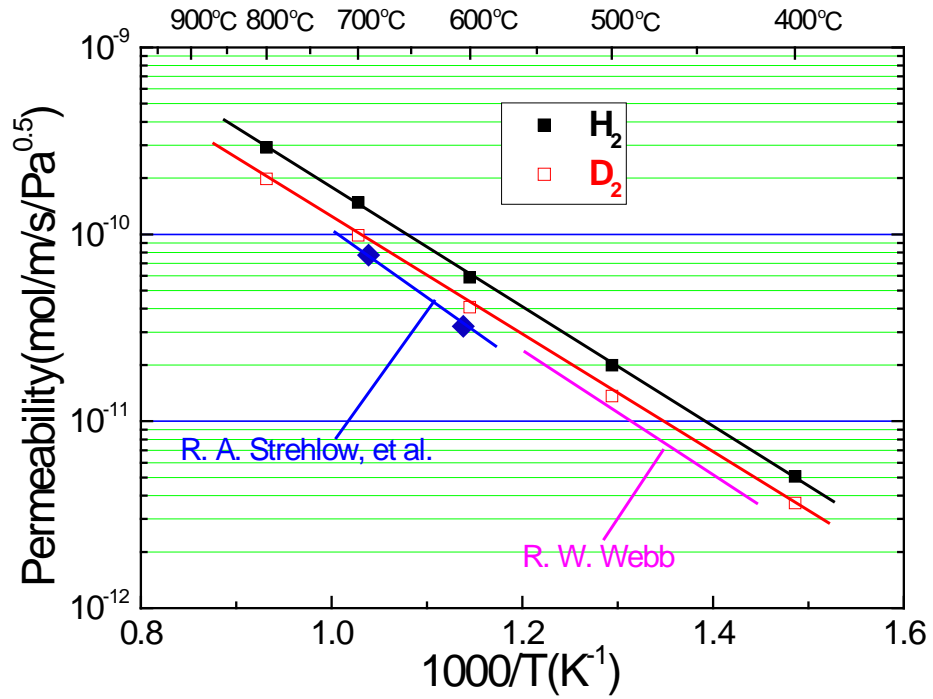


Little difference of diffusion coefficient between Deuterium and hydrogen in FLiNaK



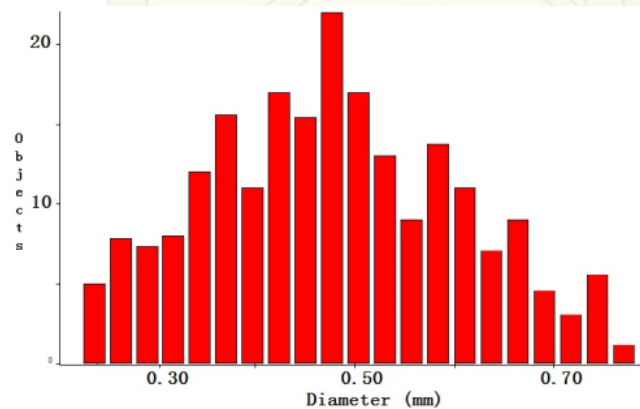
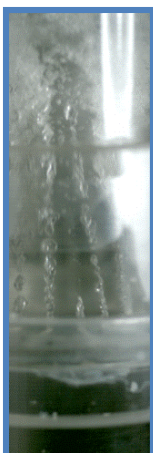
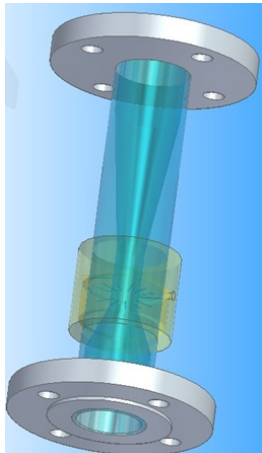
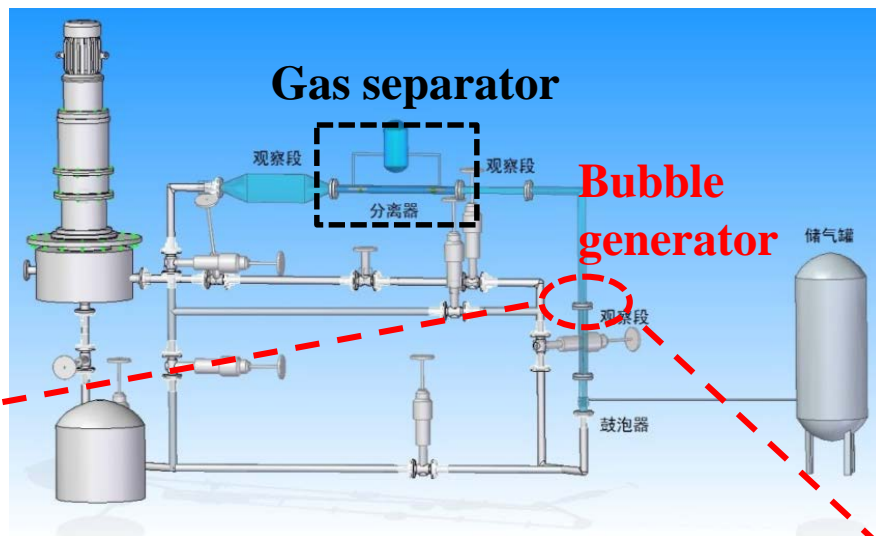
Much more solubility of deuterium than hydrogen in FLiNaK at 500-700°C

➤ Permeability of hydrogen isotopes in Hastelloy N (left) and GH 3535 (right) alloy



2. Tritium Extraction

Set up gas extraction system in water test loop and the separation efficiency was qualitatively analyzed



Average bubble diameter : 0.55mm

Bubble generator

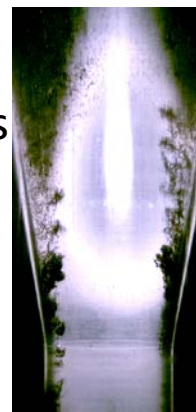
1 hole



2 holes



4 holes



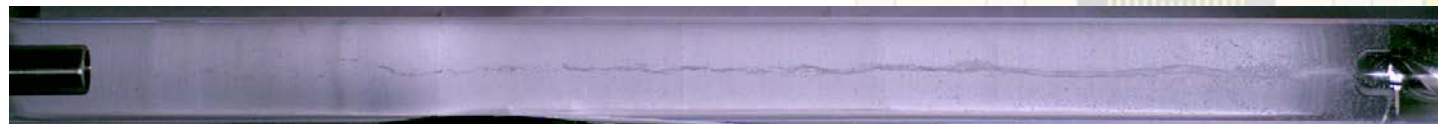
Little effect of number of holes on the bubble size distribution at the flow rate greater than $15 \text{ m}^3/\text{h}$,

Gas separator

Negative Press
0.009MPa



Balance
0.012MPa



Positive Press
0.030MPa



Separation efficiency of separator relating with the outlet pressure

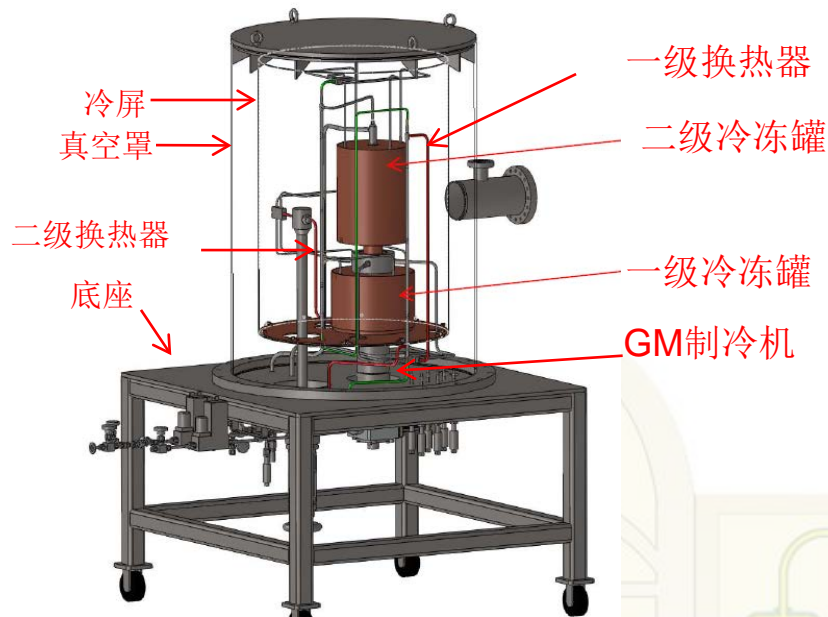
3. Tritium Separation

Successful separating xenon, krypton and hydrogen from helium environment using cryogenic technology for tritium separation

Kr、Xe <1PPb and H₂<0.1PPM after two-step frozen

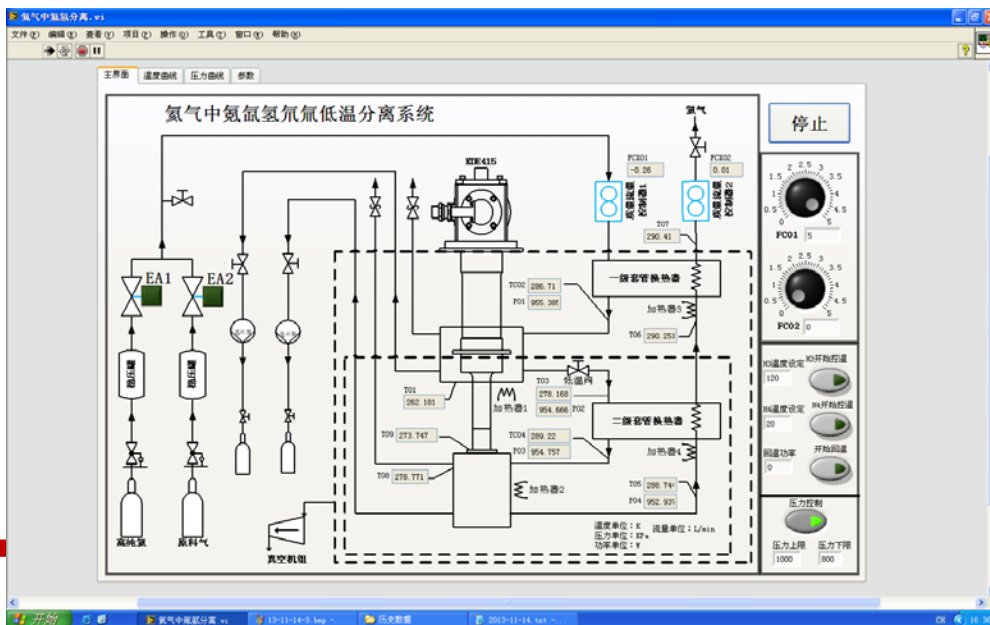
Labview interface

(In-site monitoring 9-way temperature and 4-way pressure pressures as well as data acquisition)



examination of gaseous leak

Detection of temperature and pressure



4. Tritium Storage

Complete research platform for tritium storage using alloy

Testing system for selecting tritium-storage material

(Low pressure)

(High Pressure)

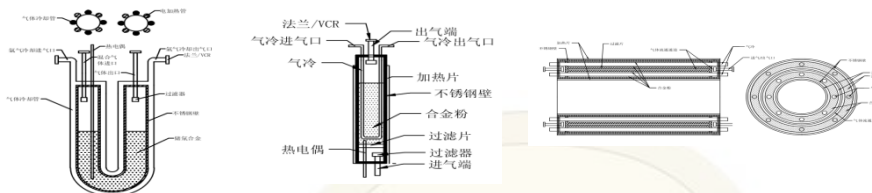


Pressure range :
0.1Pa-50KPa

Pressure range :
20KPa-10MPa

Testing system of tritium-reservoir performance

Tritium reservoir



tritium-reservoir performance **Control**

Gas distribution

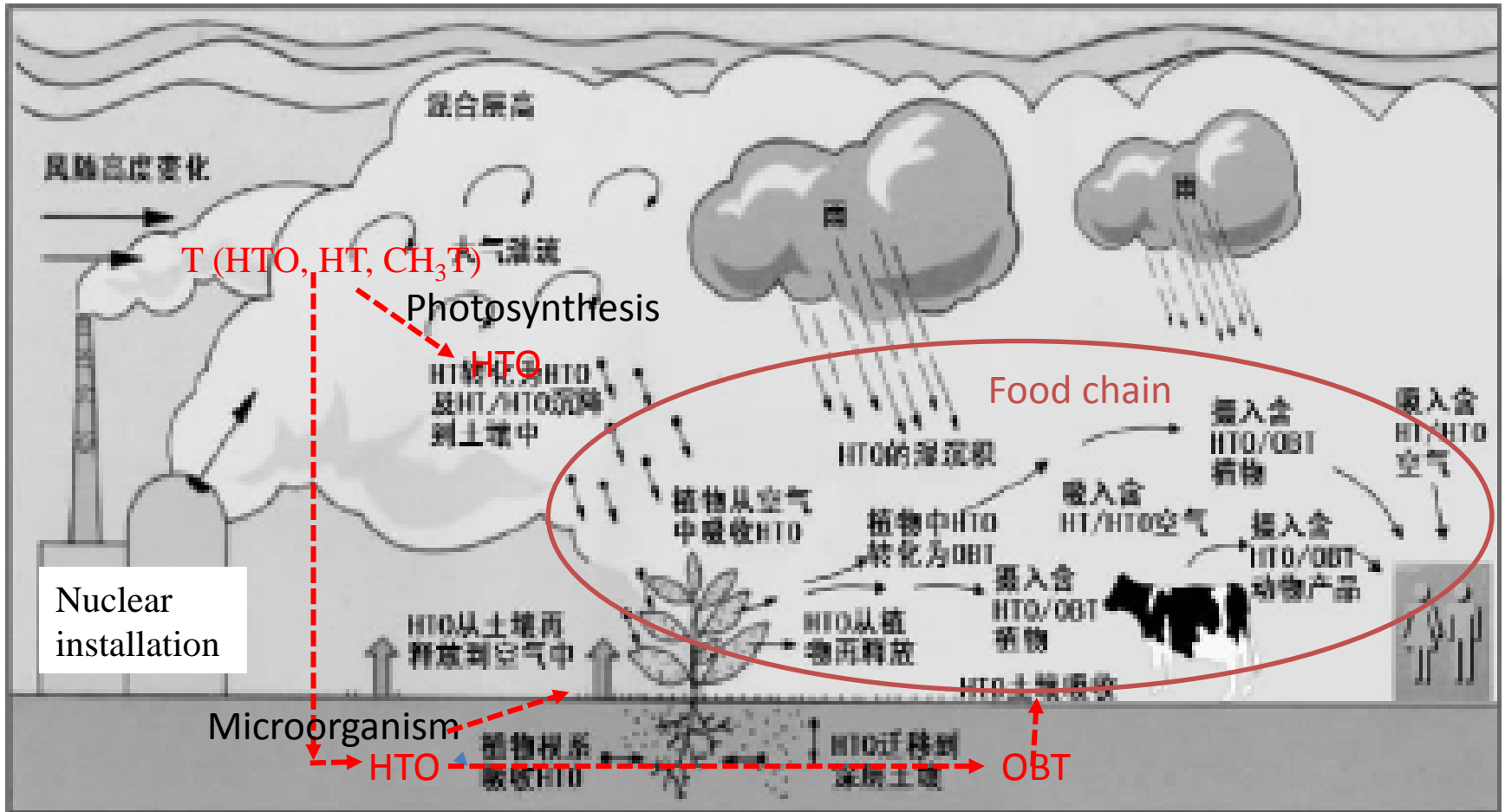


Detection



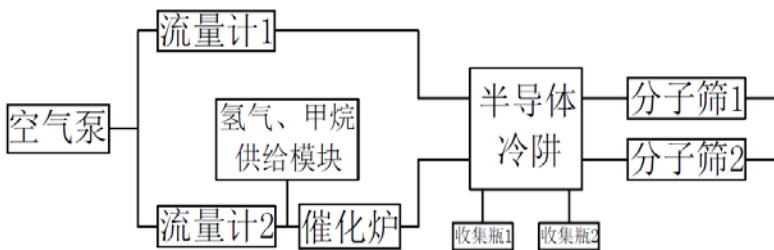
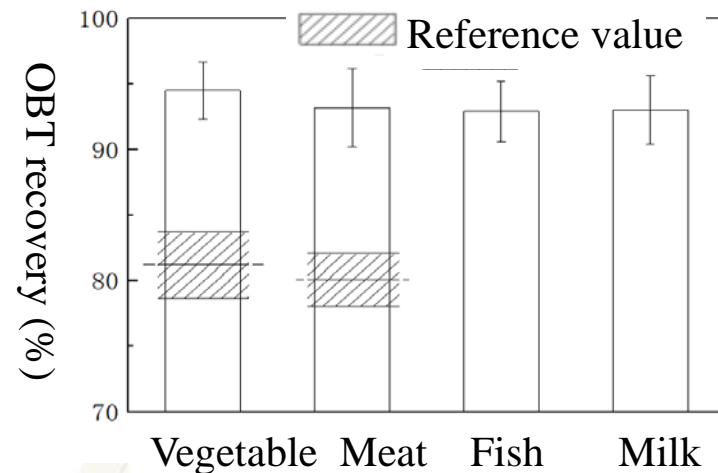
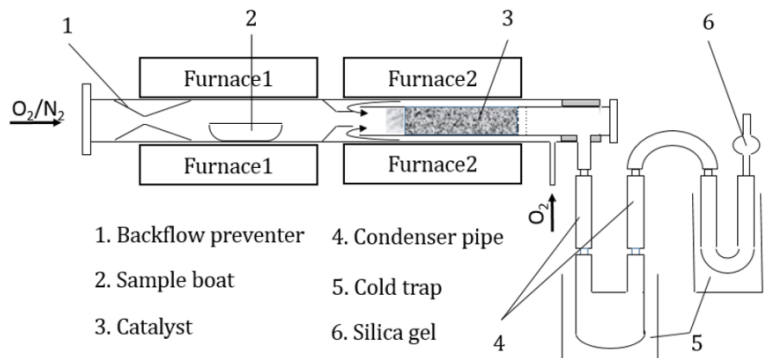
5. Tritium Sampling

Tritium transport in environment

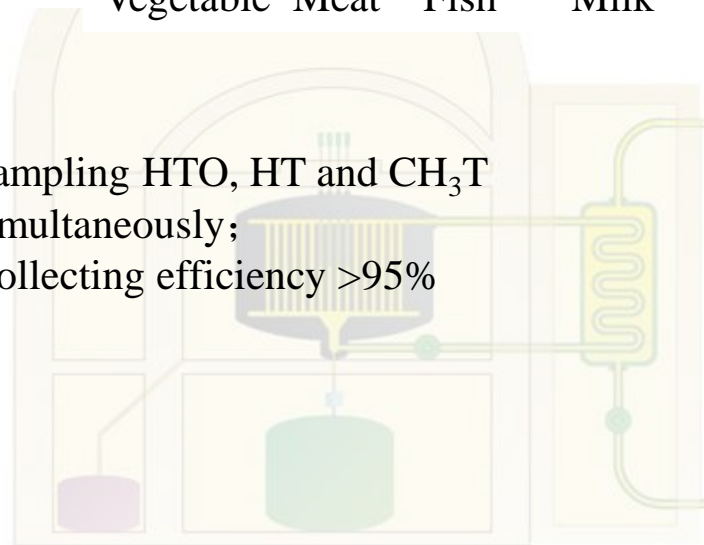


Sampling tritium (HTO, HT, CH₃T, OBT): Necessary

Successful development of OBT-oxidation collection and tritium sampler



Sampling HTO, HT and CH₃T simultaneously;
Collecting efficiency >95%



6. Tritium Monitoring

➤ Off-line Measurement

Tritium measurement platform (Detected limit of HTO < 0.5Bq/L)

Sampling

Pretreatment

Measurement

OBT collection

Tritium sampler



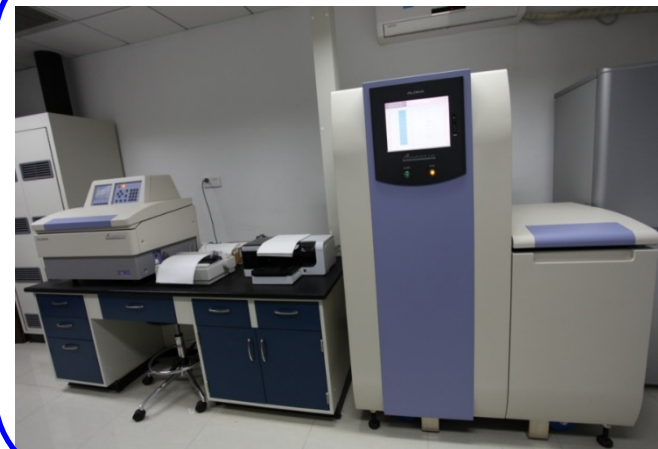
Distillation



Electrolytic enrichment



LSC



➤ Online monitoring



On line monitoring of HTO, HT and Kr, Xe, simultaneously ;

HT,HTO: $1-10^7 \mu\text{Ci}/\text{m}^3$

Kr and Xe: $1-10^6 \mu\text{Ci}/\text{m}^3$

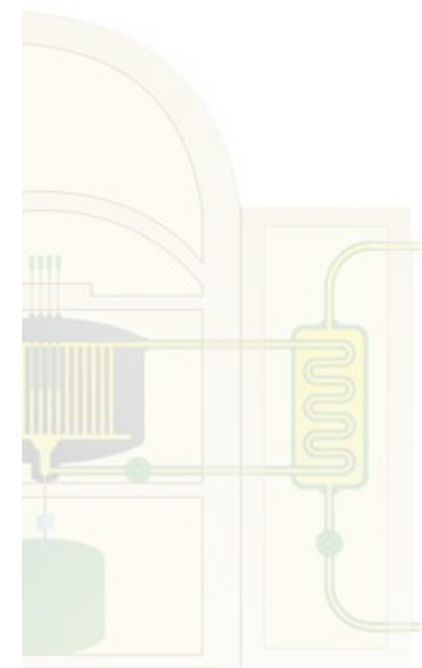
Other studies

Tritium barrier

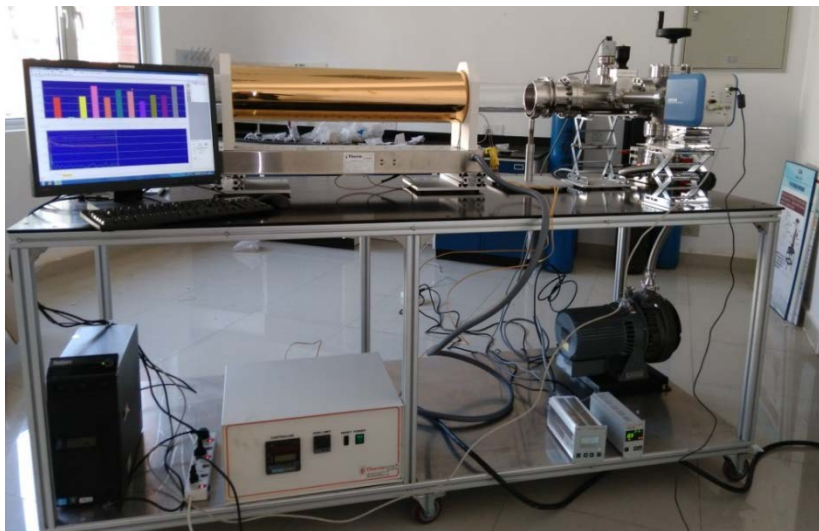
Selection of tritium barrier

Barrier	Base Metal	PRF*
Al_2O_3	SS316, MANET, TZM, Ni, Hastalloy-X	10 to >10,000
TiC, TiN, TiO_2	SS316, MANET, TZM, Ti	3 to >10,000
Cr_2O_3	SS316	10 to 100
Si	Steels	10
BN	304SS	100
N	Fe	10 to 20
Er_2O_3	Steels	40 to 700

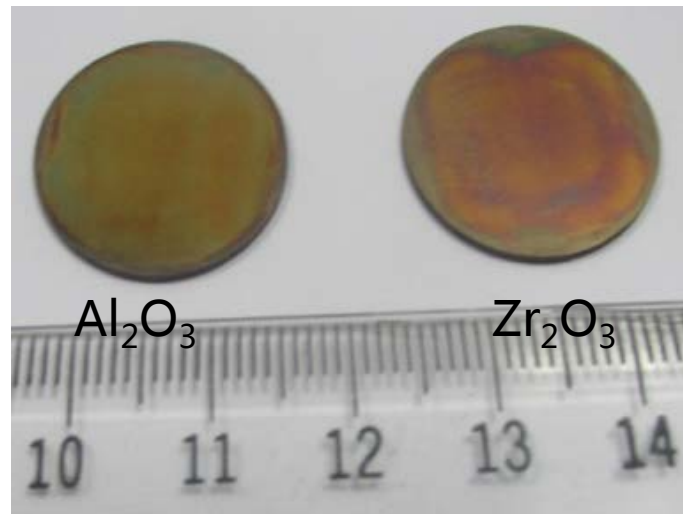
*permeation reduction factor



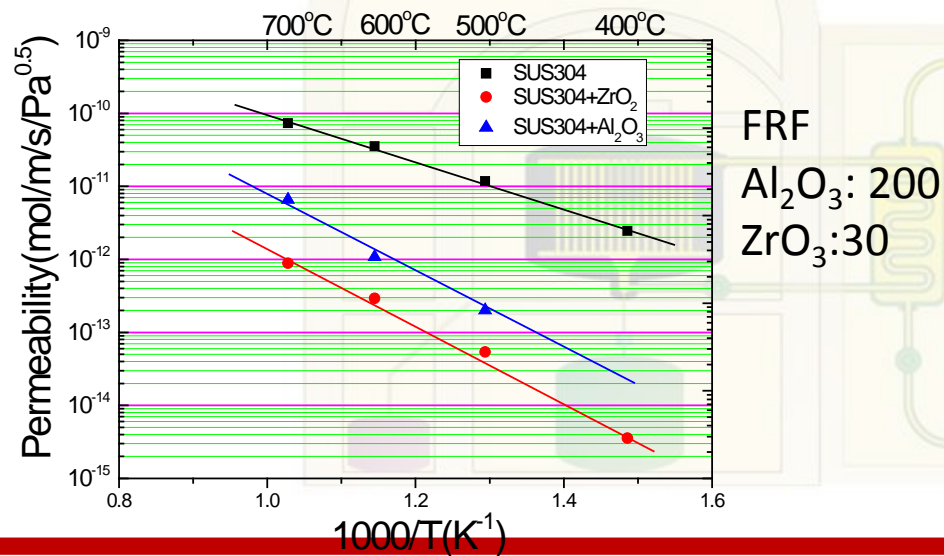
Experimental results of tritium barrier



Fabrication for tritium barrier by pack cementation



Measurement of FRF of tritium barrier



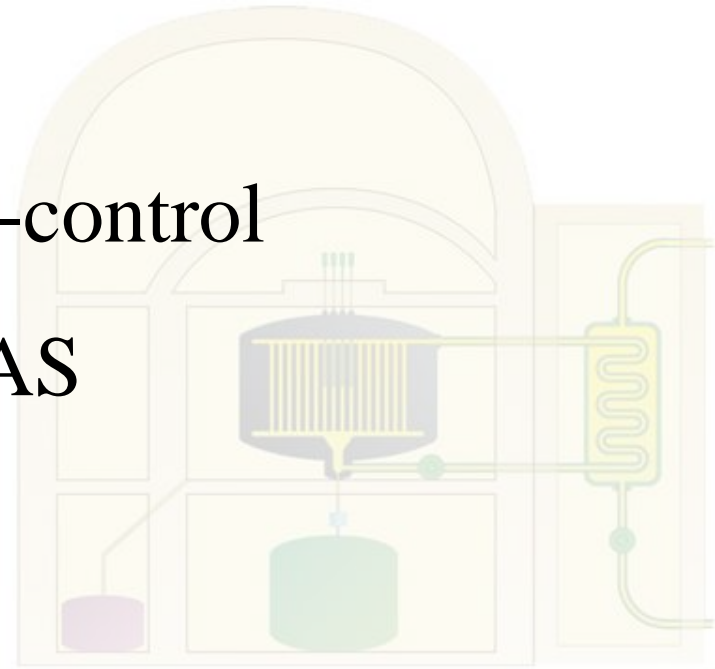
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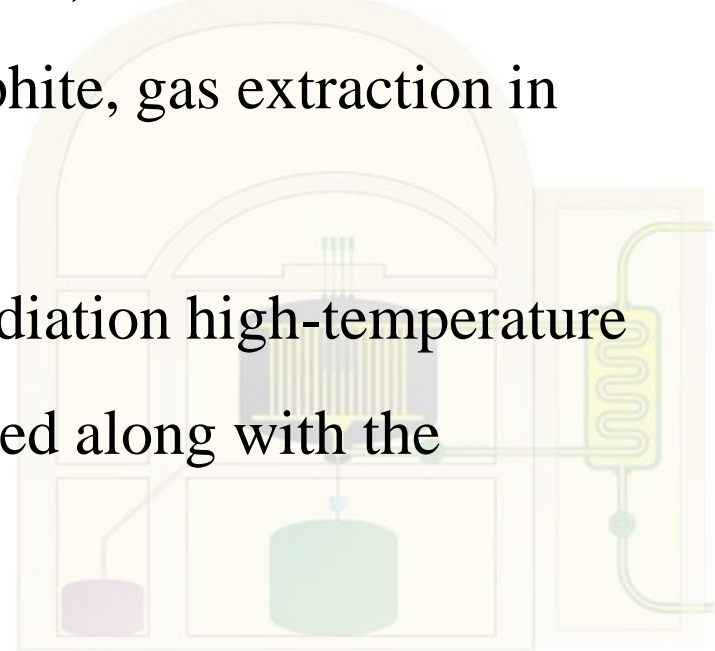
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IV. Summary

1. Several test platforms for tritium control have been set up in CAS
2. Sampling and monitoring technologies have been mastered in CAS
3. Several experiments are conducting in CAS, those consist of interaction of hydrogen isotope with graphite, gas extraction in high-temperature molten salt loop
4. The chemical form of tritium in both irradiation high-temperature molten salt and graphite will be determined along with the cooperation between SINAP and MIT





Thanks for Your Attention

